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chemical equation. a way to describe a
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Modern Chemistry Chapter 8. A number in front of a chemical formula in an equation that indicates how many molecules or atoms of each reactant and product are involved in a reaction. When the positive ion of one compound replaces the positive ion of another to form two new compounds.

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organized according to the ease with
which the elements undergo certain
chemical reactions. chemical equation.
represents, with symbols and

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formulas, the identities and relative amounts of the reactants and products in a chemical reaction.

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CHAPTER 8 REVIEW Class Science

Chemical Equations and Reactions

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SHORT ANSWER Answer the following questions in the space provided. 1. Match the symbol on the left with its appropriate description on the right. (aq) (a) (b) (c) (d) (f) A precipitate forms. A gas forms. A reversible reaction occurs. Heat is applied to the reactants.

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Describing Chemical Reactions.

OBJECTIVES ; Describe how to write
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OBJECTIVES

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CHAPTER 8 REVIEW Chemical

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Equations and Reactions SECTION 2
SHORT ANSWER Answer the following questions in the space provided.

1. Match the equation type on the left to its representation on the right.

c synthesis (a) $AX + BY \rightarrow AY + BX$ d decomposition (b) $A + BX \rightarrow AX + B$ b single-displacement (c) $A + B \rightarrow AX$ a

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double-displacement (d) $AX + AX$

8 Chemical Equations and Reactions
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