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from a kettle becomes
deposited as drops of
distilled water on a cold
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the most widely used
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principle of the
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Distillation is a process
in which a liquid
mixture of volatile
components is separated

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by imparting energy to it in consideration with the boiling points of the components so that selective vaporization takes place. This process can also be used in reverse to selectively condense the vapour mixture.

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Distillation, process involving the conversion of a liquid into vapour that is subsequently condensed back to liquid form. It is exemplified at its simplest when steam from a kettle becomes deposited as drops of

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distilled water on a cold surface. Distillation is used to

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involving the conversion
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condensed back to liquid form. It is exemplified at its simplest when steam from a kettle becomes deposited as drops of distilled water on a cold surface. Distillation is used to separate liquids from nonvolatile solids, as in the separation of alcoholic liquors from fermented materials, or in the separation of two

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or more liquids having
different boiling points,
as in the separation of
gasoline, ...

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Distillation is the most commonly used method for the separation of homogeneous liquid mixtures and is based on differences in the boiling points or relative volatility of the constituent components. 1,2 However, where the relative volatility is close

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to unity (e.g., for the separation of azeotropic mixtures or close boiling components), a third component (i.e., entrainer, solvent, or ...

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An alternative to
recover ethanol is to use
extractive distillation.

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The solvent used is
Propylene Glycol.
Recall also that ethanol
forms a minimum-
boiling azeotrope with
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Distillation refers to the selective boiling and subsequent condensation of a component in a liquid mixture. It is a separation technique that can be used to either increase the concentration of a

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particular component in the mixture or to obtain (almost) pure components from the mixture.

Distillation - Definition, Detailed Process, Types, Uses

Distillation is the process of separating the components or substances from a liquid mixture by using

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selective boiling and condensation. Distillation may result in essentially complete separation (nearly pure components), or it may be a partial separation that increases the concentration of selected components in the mixture.

Distillation - Wikipedia
Developments in

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