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2.0 mol of hydrated sodium carbonate,  $\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$ ? Avogadro's constant,  $L$  or  $N_A$   $23:6.02 \times 10^{23} \text{ mol}^{-1}$  A. 6 B. 26 24C.  $\times 103.6$  D.  $\times 101.6$  25  
2. 3What is the volume, in  $\text{cm}^3$ , of the final solution if  $100 \text{ cm}^3$  of a solution containing 1.42 g of sodium sulfate,  $\text{Na}_2\text{SO}_4$ , is diluted to the ...

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1 1 H 1.01 3 Li 6.94 11 Na 22.99 19 K 39.10  
37 Rb 85.47 55 Cs 132.91 87 Fr (223) Atomic  
number Element Relative atomic mass 25 Mn  
54.94 43 Tc 98.91 75 Re 186.21 61 Pm 146.92  
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