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answers

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reaction. A combination reaction is a chemical change in which two or more substances react to form a single new substance. As shown in Figure 11.5, magnesium metal and

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Unit 2 - Chemical Reactions

Balance the following reactions and indicate

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which of the five types of chemical reaction are being represented: 1) _____ NaBr + _____ Ca(OH) 2 _____ CaBr 2 + _____ NaOH

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the mass of the products of a chemical reaction. is always the same as the mass of the reactants. in that reaction. The atoms have just changed. partners to form new chemical bonds. 8. It is a reaction that releases energy. 9. It is a reaction that absorbs energy. 10. a. Yes. b. No. c. Yes. d. No. Section 2 (page 82) Directed Reading for ...

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Answer Key - CHEMISTRY 151 - PRECIPITATION REACTIONS KEY For each of the following reactants, predict whether a precipitation reaction will take place between them If there is no reaction, write "no reaction" If there is a reaction, write the complete, complete ionic, and net ionic equations that describe the reaction

$$2K_3PO_4(aq) + 3CdCl_2(aq)$$

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