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1) Zinc and lead (II) nitrate react to form zinc nitrate and lead. 2) Aluminum bromide and chlorine gas react to form aluminum chloride and bromine gas. 3) Sodium phosphate and calcium chloride react to form calcium phosphate and sodium chloride. 4) Potassium metal and chlorine gas combine to form potassium chloride.

## Chapter 7 Worksheet #1 Balancing Chemical Equations

Chapter-7 worksheet 1. What is the difference between cellular respiration and breathing? Cellular respiration - production of ATP energy from food and oxygen. Breathing – inhalation of oxygen and

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exhalation of carbon dioxide 2. Briefly explain oxidation and reduction reactions.

Chapter-7 worksheet answers (N).docx - Chapter-7 worksheet ...

Chapter 7 Worksheet 1. Given the following reactions, draw reasonable mechanisms that describe the chemical transformation that has taken place. To the right, list the name of the elementary mechanistic step used in the transformation. + HSO NaCl NaBH<sub>3</sub>.  
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Solved: Chapter 7 Worksheet 1. Given The Following Reactio ...

Chapter 7 Worksheet: Answers Chapter 7 Worksheet: Answers  
Practice 1: 45 Practice 2: a) Practice 3: 0.6 Practice 4: 0.9 Practice 5: 5 6  
Practice 6: 5 6 1 13 1 26 b) c) 1 2 d) 1 10 Practice 7: Event: 0 Heads 1

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Head 2 Heads 3 Heads 4 Heads Probability:  $\frac{1}{16}$   $\frac{4}{16}$   $\frac{6}{16}$   $\frac{4}{16}$   $\frac{1}{16}$

Practice 8: a) Odds for are 1:2 b) Odds against are 2:1 c) Odds on are 2:1  
Practice 9: a) 0.52 Practice 10: a)

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Chapter 7 Worksheet 1. Given the following reactions, draw reasonable mechanisms that describe the chemical transformation that has taken place. To the right, list the name of the elementary mechanistic step used in the transformation

$\text{NO}$ ,  $\text{NO}$   $\text{H}_2\text{SO}_4 + \text{HSO}_4^-$   
 $\text{NO}_2$   $\text{HSO}_4^- + \text{H}_2\text{SO}_4$   $\text{NaCl}$   $\text{NaCl}$ .

Solved: Chapter 7 Worksheet 1. Given The Following Reactio ...

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6. fresh green light. 7. joining.

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