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The Behavior of Gases

WebQuest Name _____

Period _____ Date

_____ 1. What is a gas? 2.

What does the kinetic
molecular theory

explain? 3. Describe the

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three states of matter. 4.

Describe the typical motions of atoms in a solid, a liquid and a gas.

5. The Kelvin scale is based on the concept of absolute zero.

The Behavior of Gases

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Gas Laws Worksheet atm

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$= 760.0 \text{ mm Hg} = 101.3$
 $\text{kPa} = 760.0 \text{ torr}$

Boyle ' s Law Problems:

1. If 22.5 L of nitrogen at
748 mm Hg are
compressed to 725 mm
Hg at constant
temperature.

Gas Law Webquest
Answer Key - Aplikasi
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DIRECTIONS. On this quest, you will be working in teams of 2, gathering information from each website listed. There are 2 jobs in each team: the navigator and the data recorder.

Navigator – is responsible for logging in, navigating the Internet – using the touch pad and maneuvering around the websites, and logging

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out. Data Recorder –
records all data NEATLY
in your packet
(teammate will copy
answers later).

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websites provided to find the answers to these questions about the behavior of gas.

[Filename: Student
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suggested web pages (and others you find) to learn about gases properties and behavior. Write out answers to each question using complete sentences, illustrating with diagrams or drawings whenever possible. Give the url for the web site you found most helpful for each question.

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and move, move about,

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Behavior of gases? -
Answers

“ The Behavior of
Gases ” has been ported
from Flash to HTML5.
This ChemThink tutorial
took a lot of time to

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make, and it looks great!

All of the original
controls are here, and the
question set is as
challenging as ever.

Students will adjust
temperature, number of
atoms, pressure, and
volume to see how this
affects the frequency and
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Nobel Laureate Carl Wieman, the PhET Interactive Simulations project at the University of Colorado Boulder creates free interactive math and science simulations. PhET sims are based on extensive education research and engage students through an intuitive, game-like environment where

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students learn through
exploration and
discovery.

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